1	Improvements in Thermal Efficiency and Exhaust Emissions with Ozone
2	addition in a Natural Gas-Diesel Dual Fuel Engine
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10	
10	Abstract
11	Here, ozone (O ₃) was introduced into the intake air in a natural gas fueled engine ignited
12	by diesel fuel, a natural gas - diesel dual fuel engine, to utilize the reactive O-radicals
13	decomposed from the O ₃ for the promotion of the ignition and for improvements in the
14	thermal efficiency and exhaust emissions. The engine experiments were performed over
15	a range of equivalence ratios of the natural gas in a single cylinder engine. The timing of
16	the pilot injection of the diesel fuel was varied from early in the compression stroke to
17	near top dead center to examine the changes in the effects of the O3 addition on the
18	ignition and combustion with the pilot injection timing while varying the O ₃ concentration.
19	The results showed that the combination of the O ₃ addition and the early pilot injection is
20	a means to improve the thermal efficiency and unburned emissions with a small amount

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21	of O ₃ . Furt	her, the improvement in the thermal efficiency and the reduction of the
22	unburned hy	drocarbons with the O ₃ addition are more pronounced for lower equivalence
23	ratios of nat	ural gas, while the O ₃ addition has a limited effect on the thermal efficiency
24	and the unbu	urned hydrocarbons for higher equivalence ratios of the natural gas.
25		
26	Keywords:	Dual Fuel Engine, Ozone, Natural Gas, Diesel fuel, Pilot injection, Thermal
27	Efficiency, I	Exhaust Emissions
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29	Nomenclatu	ıre:
30	CA50	Mass burned at 50% crank angle [°CA ATDC]
31	$(dp/d\theta)_{\rm max}$	Maximum rate of pressure rise [MPa/°CA]
32	$Q_{ m burn}$	Energy released with chemical reactions [J]
33	$Q_{ m cool}$	Energy lost due to cooling [J]
34	Q_{fuel}	Total fuel energy supplied [J]
35	V_{c}	Clearance volume at the top dead center [m ³]
36	$V_{ m h}$	Displacement volume [m ³]
37	$V_{ heta}$	Volume of combustion chamber at crank angle θ
38		
39	Greek syml	pols:
40	$arphi_{ m cool}$	Cooling loss [%]
41	$arphi_{ m gas}$	Equivalence ratio of natural gas [-]
42	θ	Crank angle [°CA]
43	$ heta_{ ext{pilot}}$	Pilot injection timing of diesel fuel [°CA ATDC]
44	$\eta_{ m i}$	Indicated thermal efficiency [%]
45	$\eta_{ m glh}$	Degree of constant volume heat release [%]
46	$\eta_{ m comb}$	Combustion efficiency [%]
47	$\eta_{ m u}$	Unburned loss [%]

48 **1. Introduction**

Natural gas fueled engines ignited by diesel fuel, natural gas – diesel dual fuel engines 49in the following, can help to reduce the environmental effect of internal combustion 50engines, as it may provide higher thermal efficiency with lower exhaust emissions, and 5152the methane, which is the primary constituent of natural gas, should be a candidate among 53carbon neutral fuels as it can be generated through carbon dioxide (CO₂)-free hydrogen derived from renewable energy sources and the CO₂ emitted from processes such as 54biomass power generation. 5556The drawback of such engines is that the unburned methane emissions which are released into the atmosphere have a warming potential many times that of carbon dioxide. 57The previous research reduced unburned methane emissions with optimization of the 58injection strategy of the diesel fuel and combustion chamber geometry, and by intake gas 5960 throttling [1-5]. Further improvements, however, would be needed as lean burn operation

with supercharging, which are expected to be an effective means to increase thermal
efficiency, may cause increases in the unburned methane emissions.

63 Another possible way to address the problem of increased methane release could be 64 the addition of ozone (O_3) to the intake air of engines. The O_3 addition is promising in 65 terms of rapid response and low energy consumption as O_3 can be produced with nonequilibrium plasma. The decomposition of O_3 to O-radicals occurs at temperatures around 500 K to 600 K [6-9]. Together with the O_3 decomposition, a recombination reaction of the O-radicals is initiated to reduce them to oxygen molecules (O_2), while the O-radicals react with hydrocarbons when there are coexisting hydrocarbons, enhancing the oxidation reaction [9-11].

The O₃ addition has been utilized to control the ignition timing in homogeneous 71charge compression ignition (HCCI) engines where hydrocarbons are premixed in the 72cylinder and the ignition of the hydrocarbons is enhanced by the reaction with the O-73radicals produced via the ozone decomposition [12-15]. The ignition enhancement effects $\mathbf{74}$ due to the O₃ addition have also been utilized in gasoline compression ignition (GCI) 75engines employing a two-stage direct injection strategy, in which the first injection is 76 77implemented before the O₃ decomposition to make the first injected fuel react with the O-radicals, and the ignition delay of the second injection is controlled by the temperature 78increase and the chemical species generated due to the first fuel injection [9, 16-17]. 79Nishida et al. [18] introduced O₃ into the intake air together with natural gas and 80 injected diesel fuel directly into the cylinder near top dead center. That report showed that 81 82 an O₃ concentration of 1000 ppm or more was needed to enhance the ignition [18]. This

83 O₃ concentration is considerably higher than that used to enhance the combustion of

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premixed gasoline components, that is less than 100 ppm [10, 11, 15-17], and considerable energy amounts are consumed to generate the O_3 concentration above 1000 ppm [19]. The differences in the required O_3 concentration between the gasoline components and the natural gas would suggest that the oxidation reactions of larger hydrocarbons are more likely to be affected by the O-radicals.

The purpose of the present study is to develop a means to improve the thermal 89 efficiency and exhaust emissions of natural gas-diesel dual fuel engines by utilizing the 90 ignition enhancement effects of the O₃. To achieve this, an experimental study was carried 91 out with a single cylinder engine. To examine the effects of the O₃ addition on the ignition 92of diesel fuel which consists of larger hydrocarbons, the pilot injection timing of the diesel 93 fuel was varied from the early in the compression stroke to top dead center while varying 94 95the seeded O₃ concentration. Further, the engine performance and exhaust emissions were examined over a range of engine loads to elucidate the changes in the requirement of the 96 O₃ concentration with the equivalence ratio of the natural gas. 97

98 2. Experimental Setup

99 The specifications of the test engine are detailed in Table 1. The test engine was a 100 single cylinder engine equipped with a common-rail fuel injection system. A

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101	commercially available injector (DENSO, G4S) having an 8-hole nozzle with the nominal
102	hole diameter of 0.125 mm and with the umbrella angle of 155° mounted on the cylinder
103	head. The bore and stroke are 98 mm and 110 mm, and the displacement volume is 830
104	cm ³ , the compression ratio is 17.6. A graphic of the piston is shown in Fig.1. The piston
105	has a toroidal-shaped cavity which has a diameter of 62.1 mm. The engine was connected
106	to an eddy current electric dynamometer (Tokyo Koki, W-130) to absorb the engine output.
107	The test engine has a low pressure EGR pathway and a roots type blower (OGURA
108	CLUTCH, TX04) for supercharging, but all the experiments were performed with
109	naturally aspirated engine operations here.
110	An illustration of the experimental setup is shown in Fig.2. The piezoelectric pressure
111	sensor (KISTLER, 6125B) was installed in the cylinder head, and the electric charge was
112	converted into a voltage signal with a charge amplifier (KISTLER, 5011B). The voltage
113	signal proportional to the in-cylinder pressure was recorded with a transient memory
114	board at a resolution of 0.2° CA for 150 cycles. It may be possible to examine the effect
115	of the ozone (O_3) addition on the cycle-to-cycle variation with the data recorded for 150
116	cycles. This is, however, outside the scope of the present study and will be examined in
117	the future. In the present study, cycle-averaged profiles of pressure and rate of heat release
118	are used to examine the combustion characteristics. Natural gas was delivered from a gas

cylinder into the intake pipe (diameter: 52.7 mm), located 450 mm upstream of the intake port with a mass flow controller (ALICAT, MCR-50SLPM-D). The composition of the natural gas could be varied from gas cylinder to gas cylinder, and the gas composition is detailed in Table 2, measured with a thermal conductivity detector (TCD) for one cylinder used as the reference in the present study.

Ozone (O₃) was produced with an ozonizer (EcoDesign, ED-OG-S4AD) while 124125introducing pure oxygen (O_2) from a gas cylinder, and it was delivered into the intake pipe, 450 mm upstream of the intake port. It is possible to produce O₃ from air as 126demonstrated in the previous study [9], but O₃ was made from O₂ to supply higher 127concentration of O₃ to the engine in the present study. The O₂ flow rate was measured 128with a rotameter embedded in the ozonizer. The O3 concentration was measured with a 129130UV ozone monitor (EBARA, EG-600) before the ozone-containing oxygen was introduced into the intake pipe. The volume-based ozone concentration in the intake air 131132was determined with the measured O₃ concentration and the flow rates of the oxygen and 133the intake air which was measured using a differential pressure type flowmeter. The concentrations of the total hydrocarbon (THC), carbon monoxide (CO), carbon 134135dioxide (CO₂), nitrogen oxides (NO_x), and oxygen (O₂) were measured at the intake and exhaust pipes with an exhaust gas analyzer (Best Sokki, BEX-5100D). The smoke 136

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emissions were measured with a Bosch-type smoke meter (ZEXEL, DSM-20AN). As
only insignificant smoke emissions were detected, no smoke data are presented in this
paper.

140 **3. Experimental Results and Discussion**

1413.1. Effects of the ozone addition on the combustion, engine performance, and142exhaust emissions at various pilot injection timings for a φ_{gas} of 0.42

First, the effects of the pilot injection timing of the diesel fuel on the rate of heat release (ROHR) were investigated for the fixed equivalence ratio of the natural gas, φ_{gas} , of 0.42.

The experimental conditions are detailed in Table 3. The injection quantity of the 146diesel fuel was maintained constant at the heat value of the injected diesel fuel formed 11 147HV% of the total energy supplied. The pilot injection timing, θ_{pilot} , was varied from -14814939°CA ATDC to 1°CA ATDC except where the maximum rate of pressure rise exceeded 1.2 MPa/°CA and the rotation speed could not be maintained at 1200 rpm due to misfiring. 150The O₃ concentration was increased from 0 ppm to 560 ppm unless the indicated thermal 151efficiency begun to decrease with the increase of the O₃. As the O₂ was supplied to 152generate the O₃, the intake O₂ concentration was increased to 21.5 vol.%. No EGR and 153

154 no supercharging were employed in the experiments.

Figure 3 shows the changes in the profiles of the in-cylinder pressure and the ROHR with the pilot injection timing of the diesel fuel, θ_{pilot} . The traces shown in Fig.3 (a) are for the late pilot injections, and those shown in (b) are for the early pilot injections. The number of the traces was reduced for simplicity, and the intermediate pilot injection timing, θ_{pilot} , of -19°CA ATDC was included in Fig.3 (a) and (b). Note that the combustion phase with the other pilot injection timings, not included in Fig.3, is in the order of the pilot injection timing.

At the late pilot injection timings, θ_{pilot} , of 1°CA ATDC, -4°CA ATDC, and -9°CA 162ATDC, two-stage combustion takes place: the first is a heat release by a mixture of the 163164diesel fuel and natural gas, and the second is a heat release by the natural gas which is not 165mixed well with the diesel fuel. The advance of the pilot injection timing from 1°CA ATDC to -9°CA ATDC advances the start of the first-stage combustion and increases the 166167 ROHR in the second-stage combustion. At the pilot injection timing, θ_{pilot} , of -19°CA ATDC, single-stage combustion occurs, followed by a lower ROHR due to the late 168 combustion after 10°CA ATDC. This change in the ROHR pattern from $\theta_{pilot} = -9^{\circ}CA$ 169170ATDC to $\theta_{\text{pilot}} = -19^{\circ}\text{CA}$ ATDC arises as the early pilot injection makes the diesel fuel become leaner and mixed well with the premixed natural gas. This type of combustion 171

172	can be seen in the reactivity controlled compression ignition (RCCI) strategy in which
173	the early pilot injection of more reactive fuel is implemented directly into the cylinder
174	where low reactive fuel is distributed [20-21]. Further advances of the pilot injection
175	timing from -19°CA ATDC cause retardations of the combustion phase as the diesel fuel
176	mixture becomes much leaner, something that prolongs the ignition delay and combustion
177	duration.
178	Second, the effects of the O ₃ addition were investigated for the fixed equivalence
179	ratio of the natural gas, φ_{gas} , of 0.42 while changing the O ₃ concentration and the pilot
180	injection timing, θ_{pilot} . Figure 4 shows the effects of the O ₃ concentration on the fuel mass
181	burned at the 50% crank angle, CA50. The addition of the 80 ppm O_3 can prevent the
182	misfiring caused by the early pilot injection. The CA50 advances with the increase of the
183	O3 concentration, and this is noticeable at pilot injection timings earlier than -19°CA
184	ATDC. At pilot injection timings later than -9°CA ATDC, the O ₃ addition has a smaller
185	influence on the CA50, and the higher O ₃ concentration is needed to advance the CA50
186	compared to the early pilot injection timings. This reason will be discussed in Fig.5.
187	Figure 5 shows the changes in the profiles of the in-cylinder pressure and the rate of
188	heat release (ROHR) with the O ₃ concentration. Here, the two pilot injection timings, θ_{pilot} ,
189	of (a) -4°CAATDC and (b) -34°CAATDC were chosen as late and early injection timings.

190 The traces of all the O_3 concentrations are not shown in Fig.5 for simplicity, and it was 191 confirmed that the combustion phase with the other O_3 concentrations, not included in 192 Fig.5, was in the order of the O_3 concentration.

At the late θ_{pilot} of -4°CA ATDC, the O₃ addition has a limited effect on the ROHR in 193 194the first-stage combustion, while the O₃ addition increases the ROHR in the second-stage combustion. This may be because the recombination reaction of the O-radicals which are 195196produced by the decomposition of O₃ early in the compression stroke reduces O-radical concentration until late in the compression stroke [7, 9], and the O-radicals cannot 197 enhance the first-stage combustion which is caused by the mixture of the diesel fuel and 198the natural gas. The premixed natural gas can react with the O-radicals, immediately after 199the O₃ decomposition, and the second-stage combustion which is mainly caused by the 200 201natural gas can be enhanced with O₃ of the order of hundreds of ppm.

At the early pilot injection timing, θ_{pilot} , of -34°CA ATDC, the ignition of the singlestage combustion which strongly depends on chemical reactions of the diesel fuel is enhanced with the 50 ppm O₃, and overall the combustion is advanced. This may be because the O₃ can remain undecomposed until the timing of diesel fuel injection, and the O-radicals produced from the O₃ would enhance the ignition of the diesel fuel which is mixed well with the premixed natural gas. A more detailed elucidation of the mechanism 208 of the detailed ignition processes is outside the scope of the present study, and the 209 mechanism will be examined in future work.

Figure 6 compares the effects of the O₃ addition on the THC and NO_x emissions 210between the θ_{pilot} of -4°CA ATDC and -34°CA ATDC. Without the O₃ addition (0 ppm), 211the early pilot injection at -34°CA ATDC discharges the higher ratios of THC emissions 212213than the late pilot injection at -4°CA ATDC. The O₃ addition reduces the THC emissions, 214and the reduction of the THC emissions becomes more remarkable with the early pilot injection. The primary reason of the reduction of the THC emissions may be the advanced 215combustion phase. The combustion phase is advanced with the O₃ addition for the early 216pilot injection at -34°CA ATDC as can be seen in Fig.5. At the late θ_{pilot} of -4°CA ATDC, 217the O₃ addition increases the ROHR and decreases the duration in the second-stage 218219combustion. As the cylinder pressure falls during the expansion stroke, the temperature of the unburned mixture decreases, and the unburned mixture remains unconsumed due 220to quenching of the oxidation process. 221

The nitrogen oxide (NO) formation rate depends on the flame temperature which is influenced by the local equivalence ratio in the combustion chamber. The NO formation rate changes with the combustion phase since the mixture which burns earlier in the cycle increases the peak cylinder pressure, and the higher peak cylinder pressures result in higher flame temperatures and higher NO formation rates. Further the increased residence time in high flame temperatures affects the amount of the NO formation. The late pilot injection at -4°CA ATDC forms a locally rich mixture where NO_x is generated. The NO_x emissions increase with the increase of the O₃ concentration due to the advanced combustion phase. At the θ_{pilot} of -34°CA ATDC, only low NO_x emissions are discharged due to the combustion of a leaner mixture of the diesel fuel, and the lower NO_x emissions are maintained when the O₃ concentration increases.

Figure 7 is a comparison of the effects of the ozone addition on the indicated thermal efficiency, η_i , the degree of constant volume heat release, η_{glh} , the combustion efficiency, η_{comb} , and the ratio of cooling loss, φ_{cool} , at the late and early pilot injection timings. The indicated thermal efficiency, η_i , is expressed by [22]

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$$\eta_i = \eta_{th} \eta_{glh} (1 - \eta_u - \varphi_{cool}) \tag{1}$$

Here, η_{th} is the theoretical thermal efficiency for the Otto Cycle. The unburned loss, η_{u} , is defined as $\eta_{\text{u}} = 1 - \eta_{\text{comb}}$, and the combustion efficiency, η_{comb} , can be determined with the measured exhaust emissions. The degree of constant volume heat release, η_{glh} , can be calculated by the following equation with the consideration of the profile of the apparent rate of heat release, $dQ/d\theta$:

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$$\eta_{glh} = \frac{1}{\eta_{th} \cdot Q} \int \left\{ 1 - \left(\frac{V_h + V_c}{V_\theta}\right)^{1-\kappa} \right\} \frac{dQ}{d\theta} d\theta$$
(2)

Here, κ is the ratio of specific heat, $V_{\rm h}$ is the displacement volume, $V_{\rm c}$ is the clearance 244245volume at the top dead center, and V_{θ} is the volume of the combustion chamber at crank angle θ . The ratio of cooling loss, φ_{cool} , which is defined as $\varphi_{cool} = Q_{cool}/Q_{fuel}$ where 246247 Q_{cool} is the energy lost due to cooling, and Q_{fuel} is the total fuel energy supplied, can be 248calculated with Eq.(1) as the indicated thermal efficiency, η_i , and the unburned loss, η_u , can be determined in the experiments. A similar approach for calculating the ratio of 249250cooling loss, φ_{cool} , is proposed in a reference [23] in which the ratio of cooling loss is defined as $\varphi_{cool} = Q_{cool}/Q_{burn}$ where Q_{burn} is the energy released by the chemical 251reaction, and the validity of the calculated ratio of cooling loss was assessed. 252

At the θ_{pilot} of -34°CA ATDC, the indicated thermal efficiency increases with the 253increase of the O₃ concentration, and the excess of O₃ decreases the indicated thermal 254255efficiency. This is because with the increase of the O₃ concentration, the degree of constant volume heat release, η_{glh} , and the combustion efficiency increase, η_{comb} , while 256the cooling loss, φ_{cool} , increases due to the advanced combustion phase. Further, the 257degree of constant volume heat release decreases with the excess amounts of the O₃, 258higher than 170 ppm. At the θ_{pilot} of -4°CA ATDC, the indicated thermal efficiency 259260increases with the increase in the O3 concentration, as the increase of the O3 concentration increases the degree of constant volume heat release and the combustion efficiency. With 261

the late pilot injection, the higher O_3 concentration is needed to improve the indicated thermal efficiency above that with the early pilot injection.

264Figure 8 shows the indicated thermal efficiency, η_i , CA50, the THC and NO_x 265emissions, and the O₃ concentration, at which the best indicated thermal efficiency can be achieved. The test results without the O₃ addition are shown for reference. The 266267indicated thermal efficiency is improved with the O_3 addition for all the pilot injection timings. At the θ_{pilot} of -30°CA ATDC, the indicated thermal efficiency without the O₃ 268269addition becomes close to that with the O₃ addition. Without the O₃ addition, the degree of constant volume heat release decreases due to the retardation of the CA50, and the 270combustion efficiency decreases as seen in the increase of the unburned emissions, but 271the retardation of the CA50 decreases the cooling loss. This result suggests that there is 272273an optimal CA50 at which the indicated thermal efficiency is increased.

The higher O_3 concentration is needed to achieve the best indicated thermal efficiency at the later pilot injection timings, while the best indicated thermal efficiency can be achieved at pilot injection timings earlier than -20°CA ATDC with lower O_3 concentrations. The O_3 addition reduces the THC emissions and increases the NO_x emissions for all the pilot injection timings, but the lower NO_x emissions can be achieved with the early pilot injection. Overall, the O_3 addition with the early pilot injection is advantageous in terms of the lower O₃ concentration needed to achieve higher thermal
efficiency, and low THC and NOx emissions.



of indicated thermal efficiency over a range of equivalence ratios of natural gas

Operating conditions achieving the best thermal efficiency were investigated over a range of equivalence ratios of natural gas while varying the O₃ concentration and the pilot injection timing of diesel fuel.

The experimental conditions are detailed in Table 4. The equivalence ratio of natural 287gas, φ_{gas} , was varied from 0.33 to 0.53 while the pilot injection quantity of the diesel fuel 288was fixed. Due to a failure of the injector used in section 3.1, a different injector equipped 289290with a nozzle with nominal specifications identical to the nozzle used in section 3.1 was used. There were, however, differences in the injection quantity between the injectors. As 291a result, the pilot injection quantity of the diesel fuel was slightly larger in this experiment. 292No EGR and no supercharging were employed in the experiments to examine the effects 293of the equivalence ratio without any complexity. 294

As explained in section 3.1, a higher O₃ concentration is needed to enhance the combustion at the later pilot injection timings, while the combustion is enhanced with a

297	lower O ₃ concentration at the early pilot injection timings. Considering this, the O ₃
298	concentration was varied from 0 ppm to 280 ppm at the pilot injection timings from -
299	69°CA ATDC to -29°CA ATDC, and from 0 ppm to 520 ppm at the pilot injection timings
300	from -19°CA ATDC to 1°CA ATDC. The O3 concentration was increased unless the
301	indicated thermal efficiency begun to decrease with the increase of the O ₃ . A larger flow
302	rate of O ₂ was needed to produce a higher concentration of O ₃ , however it was difficult
303	to control the O ₃ concentration precisely with the larger flow rate of O ₂ . To overcome this,
304	the flow rate of the O ₂ was changed with the pilot injection timings. As a result, the intake
305	O_2 concentration was 20.6 vol.% at the pilot injection timings from -69°CA ATDC to -
306	29°CA ATDC, and 21.5 vol.% at the pilot injection timings from -19°CA ATDC to 1°CA
307	ATDC. The differences in the intake O ₂ concentrations have a limited effect on engine
308	performance and exhaust emissions, since it was confirmed that the experimental results
309	for the various pilot injection timings show a similar tendency to that obtained in section
310	3.1.

In the following experiments, first, the engine performance and the pilot injection timings encountering the occurrence of knocking and misfiring were examined without the O₃ addition. Figure 9 compares the indicated thermal efficiency, η_i , with the equivalence ratios of the natural gas, φ_{gas} . The limit of the operations due to the misfiring 315and knocking are superimposed. The operation with the φ_{gas} of 0.33 results in a low 316 indicated thermal efficiency, and the late and early injections are limited by misfiring due to the combustion of the lean fuel-air mixture. With the higher φ_{gas} at 0.39 and 0.45, the 317318 indicated thermal efficiency increases primarily due to the improvement in the combustion efficiency, and the pilot injection without misfiring can be advanced. In 319 analogy with the trend in Fig.8, the higher indicated thermal efficiency can be achieved 320 at the earlier pilot injection timings with the φ_{gas} of 0.39 and 0.45. For the φ_{gas} of 0.53, the 321322 pilot injection timing is limited to the late crank angles due to knocking. The indicated thermal efficiency for the φ_{gas} of 0.53 is higher than the 0.39 and 0.45 φ_{gas} at the late pilot 323injection timings of -4°CA ATDC and 1°CA ATDC, while the highest thermal efficiency 324of the φ_{gas} of 0.39 and 0.45, which can be achieved at the earliest pilot injection timings, 325326 is almost equivalent to the thermal efficiency of the φ_{gas} of 0.53. Second, the O₃ was introduced, and the pilot injection timing was varied within the 327range where no knocking and misfiring occur. Figure 10 shows the effects of the O₃ 328

concentration on the fuel mass burned at the 50% crank angle, CA50. With the pilot injection timing, θ_{pilot} , later than -9°CA ATDC, the O₃ addition has a limited effect on the CA50 at all the equivalence ratios of the natural gas, φ_{gas} . For the φ_{gas} of 0.33 and 0.45, advances of the θ_{pilot} from -29°CA ATDC cause retardations of the CA50, and the CA50 is advanced with the increase of the O₃ concentration. In the present study, it was possible to operate the engine at the θ_{pilot} of -69°CA ATDC with the aid of the O₃ for the φ_{gas} at 0.33, while it was not possible to establish conditions under which the θ_{pilot} earlier than -49°CA ATDC could be set without causing misfiring and knocking for the φ_{gas} of 0.45. Further, with the θ_{pilot} earlier than -4°CA ATDC was limited by knocking with the φ_{gas} of 0.53.

339 Figure 11 shows the indicated thermal efficiency, η_i , the degree of constant volume heat release, η_{glh} , the combustion efficiency, η_{comb} , and the cooling loss, φ_{cool} , for the φ_{gas} 340 of 0.33. Without the O₃ addition, the lower indicated thermal efficiency is mainly due to 341the lower combustion efficiency that is caused by the combustion of the lean natural gas 342mixture. The O₃ addition improves the combustion efficiency and the degree of constant 343 344volume heat release noticeably with the θ_{pilot} earlier than -49°CA ATDC due to the combustion promotion effect. The excess of O₃, however, decreases the indicated thermal 345efficiency with the θ_{pilot} at -49°CA ATDC since the CA50 is advanced, and the cooling 346 loss is increased. With the θ_{pilot} at -69°CA ATDC, a simultaneous higher combustion 347efficiency and lower cooling loss is obtained with the O₃ addition because the θ_{pilot} of -348 349 69°CA ATDC allows the moderately late CA50 that reduces the cooling loss. As a result, the highest indicated thermal efficiency of 41.6% is attained at the θ_{pilot} of -69°CA ATDC 350

351	with the 140 ppm O ₃ . Figure 12 shows the indicated thermal efficiency, η_i , and the factors
352	related to the thermal efficiency for the φ_{gas} at 0.45. Unlike the φ_{gas} at 0.33, the higher
353	indicated thermal efficiency can be achieved at the θ_{pilot} of -69°CA without the O ₃ addition
354	due to the better combustion efficiency. With an increase of the O3 concentration to
355	around 50 ppm, the indicated thermal efficiency increases due to the improvements in the
356	combustion efficiency and the degree of constant volume heat release increase while
357	maintaining the lower cooling loss. The further increase of the O ₃ concentration, however,
358	increases the cooling loss. As a result, the highest indicated thermal efficiency of 41.7%
359	is attained at the θ_{pilot} of -49°CAATDC with the 30 ppm O ₃ . Figure 13 shows the indicated
360	thermal efficiency, η_i , and the factors related to the thermal efficiency for the φ_{gas} at 0.53.
361	The pilot injection timings are limited to -4°CA ATDC and 1°CA ATDC. At the late pilot
362	injection timings, the O ₃ addition slightly increases the degree of constant volume heat
363	release and the cooling loss due to the advanced CA50, and the O ₃ addition has a limited
364	effect on the indicated thermal efficiency.
365	Figure 14 shows the effect of the O ₃ concentration on the maximum rate of pressure

rise, $(dp/d\theta)_{\text{max}}$, over the range of equivalence ratios of the natural gas, φ_{gas} . The increase in the $(dp/d\theta)_{\text{max}}$ with the O₃ addition is more remarkable for the pilot injection timings, θ_{pilot} , earlier than -29°CA ATDC for (a) the φ_{gas} of 0.33 and (b) the φ_{gas} of 0.45. This is 369 because the O_3 addition advances the combustion phase of the single-stage combustion at 370 the early pilot injection, while the O₃ addition with the late pilot injection has a limited effect on the first stage combustion which determines the maximum rate of pressure rise, 371372as can be seen in Fig.5. For (c) the φ_{gas} of 0.53, the $(dp/d\theta)_{max}$ increases with the increase of the O₃ concentration at the late pilot injection of the θ_{pilot} of -9°CA ATDC as the ROHR 373374at second stage combustion becomes large and increases the maximum rate of pressure 375rise. The early pilot injections with the O₃ addition lead to the lower $(dp/d\theta)_{\text{max}}$ compared 376

to the θ_{pilot} of -9°CA ATDC for (a) the φ_{gas} of 0.33, and for (b) the φ_{gas} of 0.45, the early pilot injections with the O₃ addition cause too much advanced CA50 as seen in Fig.10 and lead to the higher $(dp/d\theta)_{\text{max}}$. This result suggests that the combination of the early pilot injection and the O₃ addition can be used with an acceptable level of the maximum rate of pressure rise at lower equivalence ratio of the natural gas while achieving an optimal CA50.

Figure 15 shows the indicated thermal efficiency, η_i , the THC and NO_x emissions, the pilot injection timing, θ_{pilot} , and the O₃ concentration, at which the best indicated thermal efficiency can be achieved. The test results without the O₃ addition are shown for reference. The improvements in the indicated thermal efficiency and the THC emissions

with the O₃ addition are most remarkable for the equivalence ratio of natural gas, φ_{gas} , of 387 388 0.33, where the θ_{pilot} is advanced to -69°CA ATDC, and the O₃ concentration is at 140 ppm. Without the aid of the O₃, it is difficult to implement the early pilot injection due to 389390 misfiring. With the increase of the φ_{gas} , the best pilot injection timing is retarded, the best O3 concentration decreases, and the differences with and without the O3 addition become 391small. For the φ_{gas} 0.53, the best indicated thermal efficiency is obtained without the O₃ 392addition at the θ_{pilot} near top dead center. There are no remarkable differences in the NO_x 393 emissions with and without the O₃ addition over a range of φ_{gas} . 394

395 4. Conclusions

To improve the thermal efficiency and exhaust emissions of natural gas fueled engines ignited by diesel fuel (natural gas – diesel dual fuel engines) with the ignition enhancement effects of O_3 , an experimental study was carried out in a single cylinder engine while varying the O_3 concentration at the intake port and the pilot injection timing of the diesel fuel. Further, the equivalence ratio of the natural gas was varied from 0.33 to 0.53 to elucidate the changes in the requirement for the O_3 concentration. The conclusions may be summarized as follows:



404		the late pilot injection. The first stage is a heat release by the mixture of the diesel fuel
405		and the natural gas, and the O ₃ addition has little influence on the rate of heat release.
406		The second stage is a heat release by the natural gas which is not mixed well with the
407		diesel fuel, and O ₃ of the order of several hundred ppm is needed to increase the rate
408		of heat release. With the early pilot injection, a single-stage heat release takes place,
409		and the ignition timing and the combustion phase are advanced with the O_3
410		concentration lower than 100 ppm.
411	2.	With the O ₃ addition, the unburned hydrocarbons decrease, the NO _x emissions
412		increase, and the early injection timing can be advanced without misfiring.
413	3.	The O ₃ concentration needed to achieve the best indicated thermal efficiency can be
414		reduced more with the early pilot injection, compared to the late pilot injection. In
415		addition, the lower unburned hydrocarbons and NO _x emissions can be achieved with
416		the combination of the O ₃ addition and the early pilot injection.
417	4.	The O ₃ addition increases the maximum rate of pressure rise notably for the early pilot
418		injection. The operation with an acceptable level of the maximum rate of pressure rise,
419		however, can be achieved at lower equivalence ratio of the natural gas while achieving
420		optimal combustion phase.
421	5.	The improvement in the indicated thermal efficiency and the reduction of the

422	unburned hydrocarbons with the O3 addition are more pronounced for lower
423	equivalence ratios of the natural gas, while the O ₃ addition has a limited effect on the
424	indicated thermal efficiency and the unburned hydrocarbons for higher equivalence
425	ratio of the natural gas. This would suggest that the combination of the O ₃ addition
426	and supercharging which reduces the equivalence ratio and decreases the cooling loss
427	is a means to improve the thermal efficiency and emissions at higher engine loads.
428	Further, the use of EGR along with the O ₃ addition may improve the indicated thermal
429	efficiency due to the decrease in the cooling loss while avoiding misfiring caused by
430	the dilution of mixtures.

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515 APPENDIX

Here, the electric energy consumption needed to produce the ozone (O₃) is calculated. The ozonizer used in the present study (EcoDesign, ED-OG-S4AD) is not manufactured specifically for the use in internal combustion engines, and its energy consumption may not be suitable for the evaluation. Instead, data of an intake in-line type ozonizer manufactured for gasoline engines, which can be found in Ref.[19], were employed to determine the energy consumption.

According to Ref.[19], the electric energy consumption of 60 W is needed to produce the O₃ concentration, [O₃], of 50 ppm with the flow rate of air at 0.2 m³/min with the intake in-line type ozonizer using dielectric-barrier discharge. For simplicity, the present calculation assumed that the electric energy consumption is proportional to the produced O₃ concentration and to the flow rate of air. Further, assuming that the volumetric efficiency of the engine is unity, the electric energy consumption, E [W], needed to produce [O₃] [ppm] is expressed as follows:

529
$$E[W] = 60[W] \times \frac{V_{\rm h} \cdot n \cdot i \, [{\rm m}^3/{\rm s}]}{0.2 \, [{\rm m}^3/{\rm s}]} \times \frac{[O_3] \, [{\rm ppm}]}{50 \, [{\rm ppm}]}$$
 (A1)

530 Here, V_h is the stroke volume [m³], *n* is the engine rotation speed [rpm], and *i* is 1/2 531 for four stroke-cycle engines.

532 The power output, N_p , [W] is expressed with the following relation:

533
$$N_{\rm p} \left[W \right] = p_{\rm me} \left[{\rm Pa} \right] \cdot \left(V_{\rm h} \cdot \frac{n}{60} \cdot i \right) \left[{\rm m}^3 / {\rm s} \right]$$
(A2)

Here, p_{me} is the mean effective pressure [Pa].

535 The ratio of the energy consumption due to the O_3 production, γ , is defined as follows:

536
$$\gamma [\%] = \frac{E}{N_{\rm p}} \times 100 = \frac{3600}{0.2 \times 50} \frac{[0_3]}{p_{\rm me}} \times 100$$
 (A3)

537 The calculated ratio of the energy consumption, γ , versus the mean effective pressure, 538 p_{me} , with the variation of the O₃ concentration is shown in Fig.A-1. The ratio of the energy 539 consumption, γ , increases with the increase of the O₃ concentration, and the energy 540 consumption is significant at the lower brake mean effective pressure.